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$\text{NH}_4\text{OH} = \text{H}^+ + \text{OH}^-$ NH_4^+ H_2S 60+ 026 0026 __ 6

D_2O $\text{Al} + \text{O}_2$ Al, O_2 + 4 H_2O K_2SO_4

(8) + BaCl_2 + KCl BaSO_4 C_2H_2 ($\text{D}_2 + \text{CO}_2 + \text{H}_2\text{O}$

$\text{Ca}(\text{OH})_2$ (N) + NH_4Cl ($\text{D}_2\text{D}^{3/4}\text{D}'\text{D}^{1/2}\text{N}$ D^1 N D° N N

$\text{D}^2\text{D}^{3/4}\text{N}$) + CaCl_2 NH_4OH) VO_5 (4) + D_2O D_2O D_2O D_2O

D_2O + $\text{Na} + \text{D}^{1/2}$; ZnI_2 ... 2th, 2024 Section 72 Types

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