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Conversely The Heat Absorbed By A System Equals The Heat Released By Its
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$C_2H_5OH(l) + 3 O_2(g) \rightarrow 2 CO_2(g) + 3 H_2O(g)$ Cot 31) A 100.0 ml. Sample Of 0.300 M NaOH Is

Mixed With A 100.0 Ml. Sample Of 0.305 M HNO_3 31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 $^{\circ}\text{C}$ And The Temperature Of The Resulting Solution Was Recorded As 37.0 $^{\circ}\text{C}$, Determine The ΔH_{rxn} (in Units Of KJ/mol). Apr 2th, 2024

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