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Show Your Work, Including Proper Units, To Earn Full Credit. 1. Silver And Nitric Acid

React According To The Following Balanced Equation: $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3$

$\text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO} \dots$ 3th, 2024

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$32.00\text{g } 1\text{mol} = 3.11\text{g O}_2$ %Yield = $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$
 $34. 4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percentpurity = 62.8% ?gpureFeCO₃ 3th, 2024

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