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Practice W/ Mol To Mol Ratios Reaction Stoichiometry Problems **Type 1: Mol To Mol Problems Example: Using The Previous Rxn, How Many Mol Of Al Can Be Produced From 13 Mol Of Aluminum Oxide? Practice W/ Mol To Mol Problems Unit Y Balanced Equation! $4Al + 3O_2 \rightarrow 2Al_2O_3$ Pg. 277, #3 - "Given" And An "Unknown" - Pg. 296 - In-class Problems 10a, 11a Mar 1th, 2024

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PRACTICE PACKET: Unit 7 Moles & Stoichiometry 5 Wwww.chempride.weebly.com ADDITIONAL PRACTICE LESSON 1: Find The Gram Formula Mass Of The Following: (Show All Work) 1. MgO 5. Ca(OH)₂ 2. NaHCO₃ 6. CH₄ 3. C₆H₁₂O₆ 7. NH₃ 4. Al May 2th, 2024

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Regents Review Stoichiometry A) NO B)NO₂ C) N₂O₃ D) N₂O₅ 25.What Is The Empirical Formula Of A Compound That Contains 30.4% Nitrogen And 69.6% Oxygen By Mass? A) SO B) SO₂ C)SO₃ D) SO₄ 26.A Compound Consists Of 40.% Sulfur And 60.% Oxygen May 4th, 2024

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Chemistry 11 Stoichiometry Review Worksheet Name: KEY 33. KClO₄ → 2O₂ + KCl Actual yield = 1.45g Theoretical yield = 5.95g KClO₄ × 1mol / 122.55g × 2O₂ 1KClO₄ × 32.00g / 1mol = 3.11g O₂ %Yield = 1.45g / 3.11g × 100% = 46.6% 34. 4FeCO₃ + O₂ → 2Fe₂O₃ + 4CO₂ Percent purity = 62.8%? g pure FeCO₃ Jun 3th, 2024

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