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Elements Make Covalent Bonds? 2th, 2024

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**Name: KEY**

Chemistry 11 Stoichiometry Review Worksheet

Name: KEY 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1\text{mol}$

$122.55\text{g} \times 2\text{O}_2 \times 1\text{KClO}_4 \times 32.00\text{g} \times 1\text{mol} = 3.11\text{g O}_2$

% Yield =  $1.45\text{g} \times 3.11\text{g} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure  $\text{FeCO}_3$  2th, 2024

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**Part #1 ...**

Review Worksheet - Nuclear Chemistry Name: \_\_\_\_\_

Part #1: Radioactivity . 1. Fill In The Table Below As A

Review. You Will Need Your Periodic Table For This!

Isotope . Total Protons (Atomic #) Total Neutrons

(Mass # - Atomic #) Mass Number Total E 4th, 2024

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AP CHEMISTRY REVIEW WORKSHEET (Unit 4 - Thermochemistry) 1.  $3 \text{ SO}_2 (\text{g}) + 2 \text{ HNO}_3 (\text{aq}) + 2 \text{ H}_2 \text{ O} (\text{l}) \rightarrow 3 \text{ H}_2 \text{ SO}_4 (\text{aq}) + 2 \text{ NO} (\text{g})$  A. Calculate  $\Delta H$  For The Above Reaction. Based On The Enthalpy Change, Determine Whether The Reaction Is Endothermic Or Exothermic And Wh 1th, 2024

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Constant, K, For The Reaction. Include Units With Your Answer. 2th, 2024

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