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Chemistry 11 Stoichiometry Review Worksheet Name: KEY 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \times 2 \times 1 \text{ KClO}_4 \times$

$32.00\text{g } 1\text{mol} = 3.11\text{g O}_2$ %Yield = $\frac{1.45\text{g } 3.11\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percentpurity = 62.8%?gpureFeCO₃ 2th, 2024

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Reaction. Based On The Enthalpy Change, Determine Whether The Reaction Is Endothermic Or Exothermic And Wh 1th, 2024

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Include Units With Your Answer. 2th, 2024

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