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Solution And State As % Of M/m, V/m, Or V/v: Given Information Identify ?/? Unit Conversion A) 10 G NaCl In 200g Of Solution B) 250 Mg NaNO₃ In 125 ML Of Solution C) 16 ML HCl In 1 L Of Solution D 2th, 2024.

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$0.791\text{g} \div 1\text{mL} \times 1\text{mol} \div 32.05\text{g} = 3.085\text{mol}$ Assuming That The Total Volume Is $125\text{ mL} + 275\text{ mL} = 400\text{ mL}$,
Molarity = Moles of Solute / Volume, 2024I. Model Problems.

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Values From A Chart. In This Case, An Approximate Value For The Tangent Of 38 Degrees Is 0.78129: $X \text{ M}$
 $X \text{ 21.876 28}(0.78129)$ (Note That We Have Included Units Of Meters, As The Original Side Was Specified In Meters.) 3th, 2024

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