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 Carbon Is Combusted In A Limited Supply Of Pure
 Oxygen. 0.50g Of The Carbon Combusts To Form CO₂
 And 0.50 G Of The Carbon Combusts To Form CO. The
 Resultant Mixture Of CO₂ And CO Is Passed Through
 Excess Na 2th, 2024.

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 to-one Molar Relationship, So 0.0322 Mol Of Aniline
 Produces 0.0322 Mol Of Acetanilide. The Amount Of
 Acetic Anhydride Does Not Play A Role, Since It Is In
 Excess. Then, N (N) 0.0322 .aniline N Aceta II. Ide Mol
 We Find The Molar Mass Of Acetanilide: (N) () 12 8 1
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