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Answers May Vary. Sample Answer: Some Advantages

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Worksheet 16 - Equilibrium Chemical Equilibrium Is The State Where The Concentrations Of All Reactants And Products Remain Constant With Time. Consider The Following Reaction:  $H_2O + CO \rightleftharpoons H_2 + CO_2$  Suppose You Were To Start The Reaction With Some Amount Of Each Reactant (and No H Mar 1th, 2024

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$F_{Pivot} = (m_B + m_1 + m_2)g$   $F_{Pivot} - m_B g - N_{B,1} - N_{B,2} = 0$  Worked Example: Solution Pivot Force: Lever Law:  $Pivot F = (m_B + m_1 + m_2)g = (2.0 \text{ Kg} + 0.3 \text{ kg}$

+0.6 Kg)(9.8 M ·s<sup>-2</sup>) =28.4 N D 1 M 1 =d 2 M 2 D2  
=d1m1 / M2 =(0.4 M)(0.3 Kg / 0.6 Kg) =0.2 M  
Generalized Lever Law , , 1 11 22, 2, ⊥ ⊥ =+ =+ FF F  
FF F & & GG G GGG Jan 2th, 2024

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A) Keq 1 D) Keq Cannot Be Determined. 6  
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