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1Geophysical Laboratory, 5251 Broad Branch Road, NW, Washington, D.C. 20015,
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Polymers In Solution

$[\eta] = KMa$ Mark-Houwink-Sakurada Equation • K, A = Constants Of Mark-Houwink Equation For Given Polymer, Solvent And Temperature; M = Molar Mass • Traditional Method Of The Determination Of Molar Mass • Viscosity Average (M V) Close To Weight-ave Mar 3th, 2024

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Weights Of 104-107 G/mol M M N W W = M Is The Mer Molecular Weight
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Angle Between The Singly Bonded Carbon Atoms Is Feb 3th, 2024

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2) Nylon Is Commonly Made By Step-growth Polymerization. In Class We Made
Nylon 6,10. A) Give The Structure Of Nylon 6,10 Circling The Chemical Group That
Defines This Polymer As Nylon. B) Draw The Structure Of The Two Chemicals That
Were Used In Class To Make Nylon 6,10. C) Explain Why This Is A Condensation
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