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Water Explain Why Ethanol ( $\text{CH}_3\text{CH}_2\text{OH}$ ) Is More

Soluble In Water Than Is Ethane ( $\text{CH}_3\text{CH}_3$ ). Answer

Ethanol Is Polar; Ethane Is Not. The Ethanol  $-\text{OH}$

Group Can Hydrogen-bond With Water. 2. Calculation

Of PH From Hydrogen Ion Concentration What Is The PH

Of A Solution That Has An H Concentration Of (a)  $1.75 \times 10^{-5}$  Mol/L;

(b)  $6.50 \times 10^{-10}$  Mol/L; (c)  $1.0 \times 10^{-4}$  Mol/L;

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