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Electrolyte. (b) A Brønsted-Lowry Acid Is A Molecule That Donates A Proton When It Encounters A Base (proton Acceptor). By This Definition,  $\text{NH}_4^+$  Can Be A Brønsted-Lowry Acid. (c) The Conjugate Acid Of A Brønsted-Lowry ... 2th, 2024  
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Chemistry 327 (Woods) Summer 2018 Page 1 Of 5  
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CHEM:3120:0003 (Tuesday, 2:00-2:50 PM In C129 PC); Or CHEM:3120:0004  
(Wednesday, 11:30-12:20 PM In C139 PC) INSTRUCTORS Primary Instructor: Prof.  
Amanda J. Haes (amanda-haes@uiowa.edu; (319) 384 - 3695) Office ... 2th, 2024.  
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