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### Molar Enthalpy Worksheet - Side A Q=m∆H

 $(Q = Energy Or Heat In KJ, M=moles, \Delta H=molar Enthalpy) A Change In Enthalpy$  $(\Delta H) Is A Measurement Of E Nergy Transfer In The Form Of Heat . M Olar Enthalpy Is$ The Enthalpy Change Per Mole Of A Substance Involved In A Transformation.Examples Of Transformations Are Phase Changes, Dissolving, And Chemical Reactions. Feb 3th, 2024

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3. (1 Mark Each) When 10.0 G Of Ammonium Nitrate Is Dissolved In 100 Cm3 Of Water, The Temperature Of The Solution Decreases From 19.0°C To 10.5°C. A Compare The Total Enthalpy Of The Reactants With That Of The Products And State Which Is Greate May 4th, 2024

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