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Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation:  $N_2 + 3H_2 \rightleftharpoons 2NH_3$  (1)  $H_0 = 92 \text{ KJ Mol}$   $S_0 = 192 \text{ J Mol K}$  3th, 2024.

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