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Clinical Approach. Philadelphia: Elsevier Health Sciences, 1996:5-17; Acta Chir Scand Suppl 1963;(suppl 306):301-65. Table 4. Electrolyte Content Of GI Fluids Fluid Average Daily Volume (mL) Sodium (mEq/L) Potassium (mEq/L) Chloride (mEq/L) Bicarbonate (mEq/L) Magnesium (mEq/L) Stomach 1000-2000 60-90 10-15 100-130 — 0.9 3th, 2024

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Normal Ranges Of Electrolytes Intracellular V Extracellular Electrolyte Composition Is Different Intracellular - K<sup>+</sup>, Mg<sup>+</sup>, PO<sup>4-</sup>, SO<sup>4-</sup>, And Proteins, And Proteins Extracellular - Na<sup>+</sup>, Ca<sup>+</sup>, Mg<sup>+</sup>, Cl<sup>-</sup>, HCO<sub>3</sub><sup>-</sup> and Lactate and Lactate Compositions Of Ions Are Maintained Selective Permeability Of Cell Membranes

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TABLE OF CONJUGATE ACID-BASE PAIRS Acid Base K A (25 OC) HClO<sub>4</sub> ClO<sub>4</sub><sup>-</sup> - H<sub>2</sub>SO<sub>4</sub> HSO<sub>4</sub><sup>-</sup> - HCl Cl<sup>-</sup> HNO<sub>3</sub> NO<sub>3</sub><sup>-</sup> - H<sub>3</sub>O<sup>+</sup> + H<sub>2</sub>O H<sub>2</sub>CrO<sub>4</sub> HCrO<sub>4</sub><sup>-</sup> - 1.8 X 10<sup>-1</sup> H<sub>2</sub>C<sub>2</sub>O<sub>4</sub> (oxalic Acid) 2th, 2024

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Chapter 3 Acid-Base Equilibria Acid-Base Equilibria Acids And Bases Play A Key Role In A Number Of Environmentally Important Chemical Reactions, Including Weathering, Transport Of Metals In Solution, And CO<sub>2</sub> Atmosphere-water Equilibria. In This Chapter We Will Develop The Concept Of An Acid And A Base, Characterize Strong And Weak Acids, 1th, 2024

### **(aq) Acid Base Conjugate Conjugate Acid Base**

Acid Base Conjugate Conjugate . Acid Base . 2) What Is The Strongest Base In The

Following Reaction?  $\text{HNO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{NO}_3^-(\text{aq}) + \text{H}_3\text{O}^+(\text{aq})$   $\text{H}_2\text{O}$  Is The Strongest Base. Strong Acids, Such As  $\text{HNO}_3$  Have Weak Conjugate Bases, So  $\text{NO}_3^-$  is A Weak Base. H 20 And 1th, 2024

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