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Binary Compound (eg. Aluminum Oxide), Use The Following Rules: 1. Write The  
Symbols Of The 2 Elements, Putting The One With The Positive Electrovalence First  
(eg. Al O) 2. Mentally Mark In The Valences Above Each Element. Jul 3th,  
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CH<sub>3</sub>CO<sub>2</sub>-Shapes Of Molecules, VSEPR • Electrons Repel • Areas Of Electron Density  
Repel Equally - Areas Of Electron Density • Single Bond • Double Bond • Triple  
Bond • Lone Pair In This Class We Are Only Concerned With The Geometry Around  
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Nitration Is The Nitro Group Can Be Reduced To An Amine! Allows The Introduction  
Of An Amine Group To The Aromatic Ring! (almost All Compounds That Contain A  
Nitrogen Attached To Aromatic Ring ! Occurred Through A Nitration)! This  
Conversion Changes The Electronic Properties Of The Ring! Nitro! Deactivating/Meta

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