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### **Section 8-3: Classifying Chemical Reactions**

Section 8-3: Classifying Chemical Reactions Reaction Type General Format Characteristics Of Reactants Characteristics Of Products Example (s) Decomposition  $AB \rightarrow A + B$  Only One Reactant Two Or More Simpler Products  $CuCO_3 (s) \rightarrow CuO + CO_2 (g)$   
 $2 H_2 O (l) \rightarrow 2 H_2 (g) + O_2 (g)$  Synthesis AKA: Combination  $O (l) + B \rightarrow AB_2$  Two Elements Or Two Simple 2th, 2024

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Specific Types Of Chemical Reactions - Synthesis, Decomposition And Combustion In Air - Will Be Explored In This Topic While The Remaining Three - Single Replacement , Double Replacement And Neutralization - Will Be Discussed 2th, 2024

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