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Thermochemical Equations - Sayre Chemistry 2Q=mct Where M Is The Mass Of The Substance C Is The Object's Specific Heat And T Is The Change In Temperature (final - Initial) Parts 2 And 4 Cannot Be Calculated With This Equation, Because There Is No Change In Temperature. Instead, They Need To Be Calculated Using Their Latent Heats Of Vaporization/fus 4th, 2024

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Solution Systems, I Have Been Using The Earlier Version Of This Table Published In 1965 In Which The Data ... 3th, 2024.
TEA: A CODE CALCULATING THERMOCHEMICAL EQUILIBRIUM ...Equation Becomes $GT/RT = R/GH(T) - H(T)/RT + 1/1000 \ln F$, 11 00 298
0 298 = $\Delta_f H^\circ(298) - \Delta_f H^\circ(0)$ Is The Fourth Term In The JANAF Tables And $F_H(298, 0)$ Is The Sixth. 1th,
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