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Calculating Lattice Energy Of BaO Crystal. Answer: The Scheme Below Represents Born-Haber Cycle For BaO Crystal. Step 1. Sublimation Of Solid Ba: $Ba(s) \rightarrow Ba(g)$,

 Δ H1 = 180.0 KJ/mol During This Process, The Change In Enthalpy Is Positive, Energy Is Required To Change The State Of Barium From Solid To Gaseous (to Break Metal Lattice). Step 2. Jun 2th, 2024

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