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1000000000000pm Or 10¹² Pm). So, Size Of The Atom Of Iron To Its Dust Particle Is $156\text{pm}/4 \times 10^6\text{pm} = 3.9 \times 10^{-5}$ Or $156\text{pm}/20 \times 10^6\text{pm} = 7.8 \times 10^{-6}$. This Means One Dust Particle Of Iron Should Be Broken Into 39million To 7.8 Million Places To Find An Iron Atom. For The Third Question, Students May Suggest That Scientists Used Microscopes To Detect The ... 3th, 2024

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