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1 2.90 M 3.18048 10 19 J 3.2 10 19 J 2. Which Is Greater, The Energy Of One Photon Of Orange 1th, 2024

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Brackets[] To Represent All The Inner Electrons , Then Just Write The Electron Configuration Of The Last Set.

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Periodic TableThe Bohr Model Of The Atom: Ground

And Excited States Ground State: The Least Energy

State An Atom Can Exist In. Excited State: When An

Atom Absorbs Energy. When The Atom Gains Energy,

The Electron Leaps To A Higher Energy Orbit. The

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SECTION 4-1 O BJECTIVES Explain The Mathematical
Relationship Among The Speed, Wavelength, And
Frequency Of Electromagnetic Radiation. Discuss The
Dual Wave-particle Nature Of Light. Discuss The
Significance Of The Photoelectric Effect And The Line-
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