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Half-life Time Elapsed Mass Left
1st 35 Min 2.50 G
2nd 70 1.25 G
3rd 105 0.625 G
4th 140 0.313 G
Rate = $k[\text{sugar}]$ And $k = 3.3 \times 10^{-4} \text{ Sec}^{-1}$. Half-life. Half-life Is 35 Min. Start With 5.00 G 4th, 2024Chemical Reactivity Worksheet • UW Chemistry

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Probing Polymer Effects On Chemical Reactivity: Ligand ... $1.34 \text{ M}^{-1} \text{ s}^{-1}$, $\Delta H^\ddagger = 11.75 \text{ Kcal/mol}$. Methanol Was Used To Dissolve 2-N⁹, The Value Quoted Is Calculated By Extrapolation To 0% Methanol From 2th, 2024.
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