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Chapter 7 Chemical Reactions Section 7.1 Describing Reactions (pages 192–198) This Section Discusses The Use Of Chemical Equations And How To Balance Them. It

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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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