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X Lab Data . 1. Measure Total # KJ (calorimetry), Using $Q = C \dots$ Hess's Law . The Total Enthalpy Change For A Chemical Or Physical Change Is The Same Whether It Takes Place In One Or Several Steps. • "state Function" Version 1: A 1th, 2024

Thermochemistry, Thermodynamics Thermochemistry

Standard States The Thermodynamic Standard State Of A Substance Is Its Most Stable Pure Form Under Standard Pressure (1 Bar) And Some Speci C Temperature (298 K Unless Otherwise Speci Ed). I For A Pure Substance In The Liquid Or Solid Phase, The Standard State Is Th 3th, 2024

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Section 17.1 - The Flow Of Energy Energy Is The
Capacity To Do Work Or Supply Heat. Energy Is The
Capacity To Do Work Or Supply Heat. Energy Has No
Mass Or Volume. Energy Has No Mass Or Volume.
Chemical Potential Energy Is Energy Stored In
Chemicals. 4th, 2024

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$C_2H_5OH(l) + 3 O_2(g) \rightarrow 2 CO_2(g) + 3 H_2O(g)$ $\Delta H_{rxn} = -1366.8 \text{ kJ/mol}$
A 100.0 mL Sample Of 0.300 M NaOH Is Mixed With A 100.0 mL
Sample Of 0.305 M HNO_3 In A Coffee Cup
Calorimeter. If Both Solutions Were Initially At 35.0 °C
And The Temperature Of The Resulting Solution Was
Recorded As 37.0 °C, Determine The ΔH_{rxn} (in Units Of
kJ/mol). 4th, 2024

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Systems," And IAR Bulletin 114, "Identification Of
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A)endothermic, Positive 4th, 2024

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