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 $+ (2.66 \times 10^{-23}) = 2.994 \times 10^{-23}$ (I Have
Given The Answer To The Same Degree Of Accuracy As
The Data We Were Given) B) The Above Is The Mass Of
1 Molecule Of Water. To Find Out How Many Molecules
There Are In 1 Gram We Divide 1 By The Mass Of 1
Molecule. $1 \div (2.994 \times 10^{-23}) = 3.34 \times 10^{22}$ Question
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Answer Mark Comments 7(a) $20(5)^3 + K(5)^2 + 9(5) -$

$= 0$ M1 $125 + 25K = 45 - 200$ A1 $K = -6$ A1 7(b) X^3

$M1 -6x^2 + 9x - 20 - (x^2 + Kx + 4)$ Sight Of X^2

And 4 In A Quadratic Factor $(x - 5)(2x + 4)$ A17c

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