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Neutral) When Equal Volumes Of The Following Are Mixed. Explain.  $0.5 \text{ M NaOH} + 0.5 \text{ M HC}_2\text{H}_3\text{O}_2$  The Following Reaction Occurs:  $\text{HC}_2\text{H}_3\text{O}_2 + \text{OH}^- \rightarrow \text{C}_2\text{H}_3\text{O}_2^- + \text{H}_2\text{O}$  - Since The Strong Base Exactly Neutralizes Th Jun 2th, 2024.

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