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** Acids: Start With "H" Or End In $-\text{CO}_2\text{H}$ ($-\text{COOH}$) And Have A Sour Taste. ** Bases: End In $-\text{OH}$ (positive Ion In Front) Or End In $-\text{NH}$, $-\text{NH}_2$, NH_3 . They Have A Bitter Taste And Feel Slippery. * 3 Acid/Base Theories . I. Arrhenius Theory A) Acids Produce H^+ Jul 1th, 2024.

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Acids And Bases
Chapter 15: Acids And Bases Arrhenius Definitions: ... In Chapter 4 We Defined Weak Acids And Weak Bases As Weak Electrolytes (only Partially Ionized In Aqueous Solution). ... Calc Apr 2th, 2024
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