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$Q\Delta t$  Where  $\Delta t$  = Difference Between Entering And Leaving Dry-bulb Temperature And  $Q$  = Airflow Rate In Cubic Feet Per Minute Latent Heat (Btu/h) =  $0.68 Q\Delta g$  Where  $\Delta g$  = Difference In Moisture Content Of Entering And Leaving Air, Grains Per Pound Of Dry Air Water May 1th, 2024

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23 0.0435 12 0.174 16 0.00726 0.003625 0.003625 0.003625 0.01088 0.003625 Empirical Formula =  $\text{Na}_2\text{CO}_3$  2 1 3 5) 0.150 G Of Copper Reacts With Oxygen Form 0.188 G Of Copper Oxide. Find The Empirical Formula Of Copper Oxide. Mass Cu = 0.150 G Cu O Mass O = 0.188 - 0.150 = 0.038 G 0.150 63.5 0.038 16 0.00236 0.00236 0.002375 0.00236 Mar 3th, 2024

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C U M F E R E N C E Radius Centre Diameter B A Adj C Opp Hyp Xo W L B B H A H H Compound Measures Speed = Density = Pressure = Volumes Cuboid =  $L \times W \times H$  Prism = Area Of Cross Section  $\times$  Length Cylinder =  $\pi r^2 h$  Volume Of Pyramid =  $\frac{1}{3} \times$  Area Of Base  $\times$  H Distance Time Mass Volume Force Area Speed Density Pres Apr 1th, 2024

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