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(d) The Oxidation States Of Vanadium In Its Compounds Are V(+5), V(+4), V(+3) And V(+2). The Vanadium(III) Ion Can Behave As A Reductant Or An Oxidant. (i) Indicate On The Following Equation Which Reactant Is The Oxidant.  $2V^{3+} + Zn \rightarrow 2V^{2+} + Zn^{2+}$  [1] (ii) Which Change In The Following Jun 1th, 2024

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(d) Molten Zinc Chloride Can Be Electrolysed Using The Apparatus Shown Below. +- A C D B (i) Which One Of The Letters, A, B, C Or D, Represents The Cathode?..... [1] (ii) Which One Of The Following Substances Is The Most Suitable For Use As An Electrode In Mar 3th, 2024

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(iii) Copper Can Be Extracted By The Electrolysis Of An Aqueous Solution. Suggest Why The Electrolysis Of An Aqueous Solution Cannot Be Used To Extract Aluminium..... [2] 2-(b) The Ions Which Are Involved In The Electrolysis Are  $Al^{3+}$  And  $O^{2-}$ . The Products Of This Electrolysis Are Given On ... May 2th, 2024

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Explain How Aqueous Sodium Hydroxide Can Be Used To Separate ... Alkenes Can Be Oxidised To Carboxylic Acids By Boiling With Aqueous Potassium Manganate(VII). (i) Pent-2-ene,  $CH_3-CH_2-CH=CH-CH_3$ , Oxidises To  $CH_3-CH_2-COOH$  And  $CH_3-CH_2-COOH$  ... 79 Se Selenium 34 80 Br Bromine 35 84 Kr Krypton 36 39 K Potassium 19 88 Sr Strontium 38 89 Y ... Jun 2th, 2024

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38 Which Statement About A Family Of Organic Compounds Describes An Homologous Series? All Compounds In The Family Have The Same A Functional Group. B Physical Properties. C Relative Molecular Mass. D Structural Formula. 39

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